



## Synthesis of (–)- and (+)-hyrtiosal and their C-16 epimers

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**Abstract**—A total synthesis of the marine sponge metabolite (–)-hyrtiosal along with its enantiomer (+)-hyrtiosal and their C-16 epimers were achieved starting from 30% ee copalic acid. © 2002 Elsevier Science Ltd. All rights reserved.

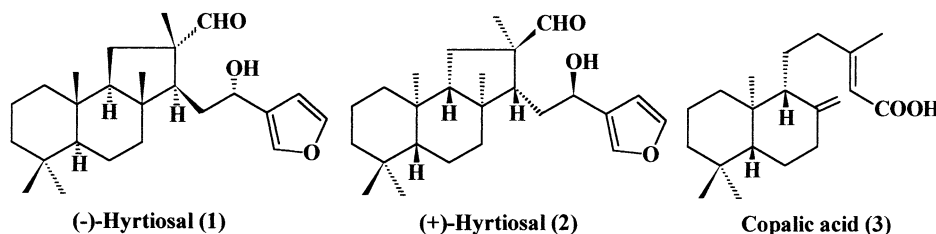
Marine sponges are a rich source of new marine natural products and the majority of compounds are reported focusing on biomimetic or ecological importance.<sup>1</sup> Many biologically active metabolites are terpenoids and (–)-hyrtiosal (**1**) is an example of sesterterpenoid isolated from the Okinawan marine sponge *Hyrtios erectus* which showed a potent cytotoxic activity against KB cell in vitro.<sup>2</sup> Its structure was elucidated spectroscopically and by chemical reactions and consists of a novel rearranged tricyclic skeleton named hyrtiosane. The absolute configuration of the natural product was recently established by total synthesis.<sup>3</sup>

Here we report the synthesis of (–)-hyrtiosal (**1**), (+)-hyrtiosal (**2**) and their C-16 epimers, starting from 30% ee copalic acid (**3**)<sup>4</sup> (Fig. 1).

Isocopalenol (**4**), synthesized previously from **3** in three steps,<sup>5</sup> was mesylated in pyridine to give **5** in 95% yield (Scheme 1). The reaction of **5** with sodium cyanide in the presence of Adogen 464<sup>®</sup> gave nitrile **6** in 70% yield along with an elimination product (ca. 20%). Reduction of **6** with DIBAL in toluene, gave aldehyde **7** in good yield (87%). Treatment of **7** with 3-bromofuran and *n*-BuLi gives a (1.2:1) mixture of epimeric

alcohols **8** (48.5%) and **9** (40.5%) that was separated by silica gel column chromatography. Since the purity of the starting copalic acid (**3**) was ~30% ee, esterification of **8** and **9** with (*S*)-*O*-methyl mandelic acid seems to be suitable for resolution and determination of the stereogenic center at C-16.<sup>7</sup> Thus, treatment of **8** with (*S*)-*O*-methyl mandelic acid in the presence of DCC and DMAP furnished two diastereoisomeric esters **10** (52%) and **11** (28%), easily separated by column chromatography (Scheme 2). Using the same sequence, **9** furnished compounds **12** (54%) and **13** (29%), separated by HPLC. Thus, the stereochemistry at C-16 of 4 stereoisomers could be established as shown in Scheme 2, according to Trost's model.<sup>7</sup>

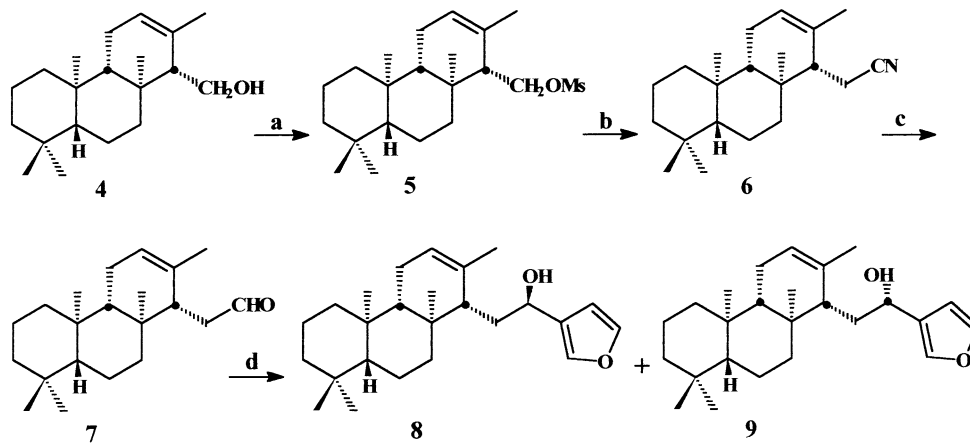
Epoxidation of **11** with *m*CPBA at –40°C furnished **14** in 70% yield and subsequent rearrangement of the epoxide with BF<sub>3</sub>·Et<sub>2</sub>O at –23°C in nitromethane gave **15** in 78% yield (Scheme 3). Finally, hydrolysis of ester (*S*)-*O*-methyl mandelate was performed with K<sub>2</sub>CO<sub>3</sub> in MeOH:THF:H<sub>2</sub>O (1:1:0.5) to give (–)-hyrtiosal (**1**) {[α]<sub>D</sub><sup>20</sup> = –72.1° (c 0.59, CHCl<sub>3</sub>)} in 66% yield. Literature<sup>2,3</sup> {[α]<sub>D</sub><sup>20</sup> = –73.8° (c 0.42, CHCl<sub>3</sub>)} and {[α]<sub>D</sub><sup>20</sup> = –62.2° (c 0.74, CHCl<sub>3</sub>)}, respectively.



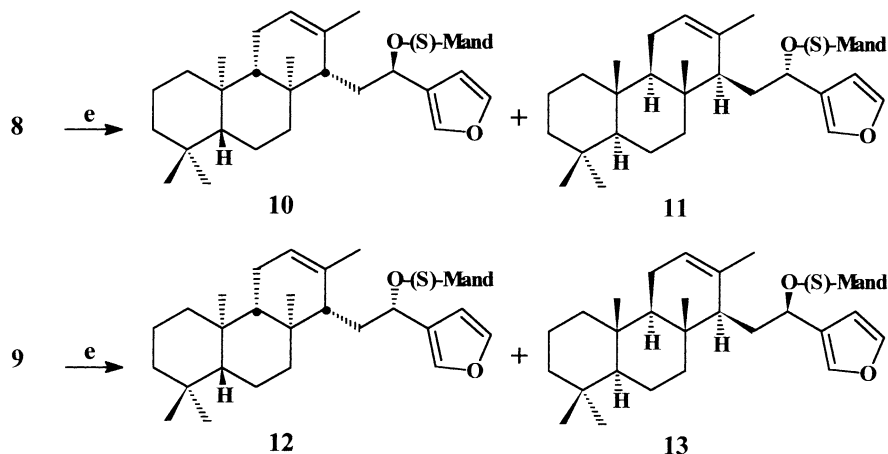
**Figure 1.**

**Keywords:** (+)-hyrtiosal; (–)-hyrtiosal; sesterterpene; copalic acid.

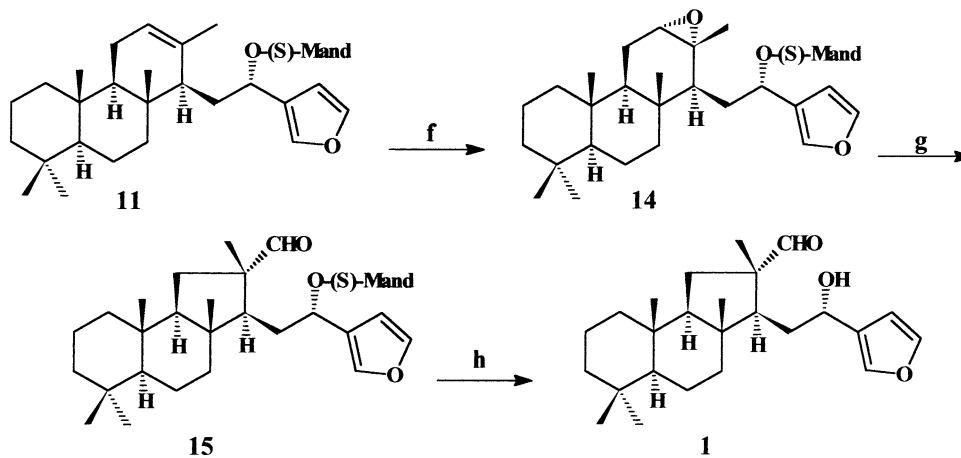
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**Scheme 1.** Reagents, conditions and yields: (a) MsCl, pyridine, rt, 15 h, 95%; (b) NaCN, Adogen 464<sup>®</sup>, toluene:H<sub>2</sub>O (10:1), 60°C, 36 h, 70%; (c) DIBAL, toluene, 0°C, 1.5 h, 87%; (d) *n*-BuLi/3-bromofuran/THF, -78°C, 1.5 h, **8** (48.5%) and **9** (40.5%)



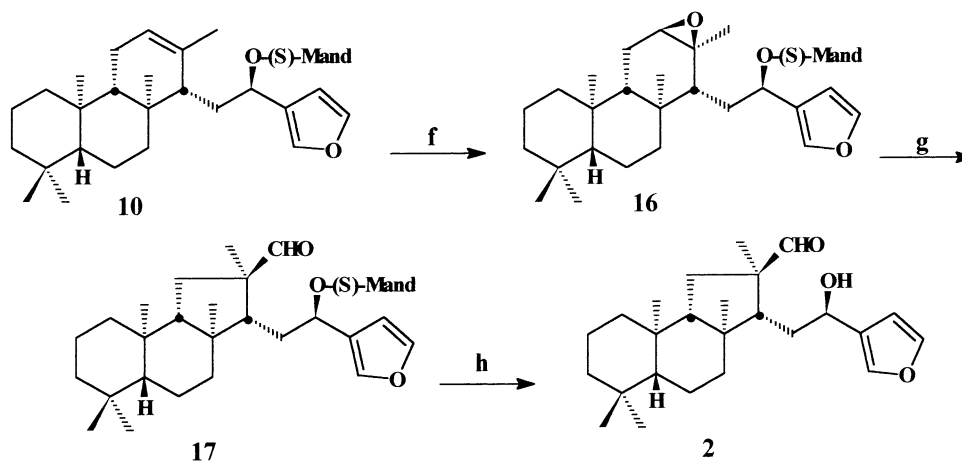
**Scheme 2.** Reagents, conditions and yields: (e) (*S*)-(+)-*O*-methyl mandelic acid, DCC, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, rt, 2 h, **10** (52%), **11** (28%), **12** (54%) and **13** (29%).



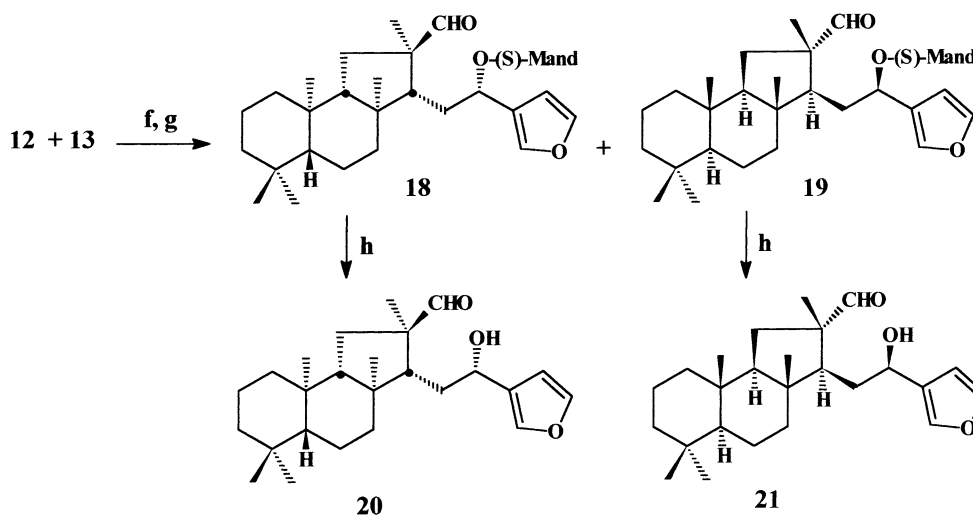
**Scheme 3.** Reagents, conditions and yields: (f) *m*CPBA, CH<sub>2</sub>Cl<sub>2</sub>, -40°C, 1.5 h, 70%; (g) BF<sub>3</sub>·Et<sub>2</sub>O, CH<sub>3</sub>NO<sub>2</sub>, -23°C, 40 min, 78%; (h) K<sub>2</sub>CO<sub>3</sub>, MeOH:THF:H<sub>2</sub>O (1:1:0.5), rt, 16 h, 66%.

Following the same sequence for the synthesis of **1** from **11**, compound **10** gave, after epoxidation, rearrangement of epoxide and hydrolysis of ester (*S*)-*O*-methyl mandelate, (+)-hyrtiosal (**2**)  $\{[\alpha]_D^{20} = +64.2^\circ$  (*c* 0.53, CHCl<sub>3</sub>) $\}$  in 31% yield for three steps (Scheme 4).

In another sequence, although the separation of **12** and **13** could be performed at this stage, the mixture was epoxidized and submitted to the acidic rearrangements to give the aldehydes **18** and **19**, which were promptly separated by HPLC (Scheme 5).



**Scheme 4.** Reagents, conditions and yields: (f) *m*CPBA, CH<sub>2</sub>Cl<sub>2</sub>, –40°C, 1.5 h, 71%; (g) BF<sub>3</sub>·Et<sub>2</sub>O, CH<sub>3</sub>NO<sub>2</sub>, –23°C, 1 h, 67%; (h) K<sub>2</sub>CO<sub>3</sub>, MeOH:THF:H<sub>2</sub>O (1:1:0.5), rt, 16 h, 66%.



**Scheme 5.** Reagents, conditions and yields: (f) *m*CPBA, CH<sub>2</sub>Cl<sub>2</sub>, –40°C, 1.5 h, 88%; (g) BF<sub>3</sub>·Et<sub>2</sub>O, CH<sub>3</sub>NO<sub>2</sub>, –23°C, 1 h, **18** (49.5%), **19** (33.5%); (h) K<sub>2</sub>CO<sub>3</sub>, MeOH:THF:H<sub>2</sub>O (1:1:0.5), rt, 16 h, **20** (69%), **21** (75%).

Finally, the hydrolysis of esters **18** and **19**, separately, furnished compounds **20** (69%) and **21** (75%), respectively, the corresponding C-16 epimers of **2** and **1**. In summary, starting from 30% ee copalic acid, we completed the syntheses of (–)-hyrtiosal, (+)-hyrtiosal and their C-16 epimers.

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#### References

1. Faulkner, D. J. *Nat. Prod. Rep.* **2000**, *17*, 7–55 and references cited therein.
2. Iguchi, K.; Shimada, Y.; Yamada, Y. *J. Org. Chem.* **1992**, *57*, 522–524.
3. Basabe, P.; Diego, A.; Diez, D.; Marcos, I. S.; Urones, J. G. *Synlett* **2000**, 1807–1809.
4. The enantiomer excess (ee) of copalic acid depends on the source of the copaiba oil in the market. See: Pantarotto, H.; Imamura, P. M. *Liebigs Ann. Chim.* **1995**, 1891–1894.
5. Imamura, P. M.; Rúveda, E. A. *J. Org. Chem.* **1980**, *45*, 510–515.
6. Heissler, D.; Ladenburger, C. *Tetrahedron* **1988**, *44*, 2513–2521.
7. Trost, B. M.; Belletire, J. L.; Goldleski, S.; McDougal, P. G.; Balkovec, J. M. *J. Org. Chem.* **1986**, *51*, 2370–2374.